

CAN RADIOMETRIC DATING BE INACCURATE? BY JOE SPEARS

Dating methods based on radioactivity have been used to indicate a vast age, in the millions or billions of years, for certain rocks. However, in some cases, other methods yield much younger ages. For example, some zircons have been radiometrically dated to about four billion years old. The decay process for the zircons involves alpha particle emission from uranium and/or thorium inside the zircon. An alpha particle is simply a helium nucleus, which will quickly gather up two electrons and become a helium atom. The problem with dating zircons is that the amount of helium in the zircon, presumably from alpha particle generation indicates an age of only a few thousand years.¹

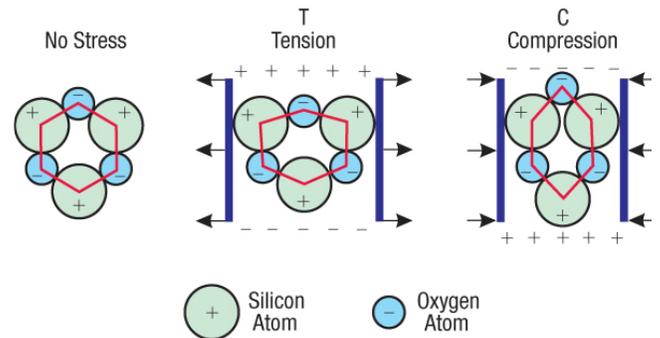
Why do we observe such disparate dates using these two different dating methods? Let's look at how the radiometric dating method works. Over time, a certain amount of radioactive material decays. For example, some isotopes of uranium can decay into lead. The process involves several intermediate steps: element *a* decays into element *b* which then decays into element *c*, etc. And there is a decay rate for each step. The decay rate is often described by the half-life of the radioactive material. This is the amount of time it takes for half of the radioactive material to decay, so after a period of time equal to the half-life, half of the radioactive material would be left in the sample. If we waited the same length of time again, we would then have a fourth (half of a half) as much radioactive material as we started with. Another half-life would leave us with an eighth of the material, and so forth. By measuring the amount of radioactive material left in a sample, and assuming we also know the amount of radioactive material that was in the sample originally, we can theoretically deduce the amount of time that was required for the original amount to decay down to the currently measured amount.

Various radiometric dating methods have one thing in common—they assume a constant radioactive decay rate (of the material being used in the dating). Is this assumption valid? Can the radioactive decay rate be changed?

¹ Humphreys DR (2005) Young Helium Diffusion Age of Zircons Supports Accelerated Nuclear Decay, *Radioisotopes and the Age of the Earth*, Institute for Creation Research, El Cajon, CA, 25-100

Figure 1

Piezoelectric Effect in Quartz



Before answering this, consider this question: What happens in nuclear reactors? Why, radioactive decay happens. And this decay is managed—control rods are used to speed or slow the reaction. So, the answer to the first question is, “Yes, radioactive decay rates can indeed be changed and are actually changed in practice.”

But the reactor is an artificial environment. Does the decay rate of radioactive materials ever change in a natural environment? What conditions would be required in nature to change the radioactive decay rate? Did those conditions ever exist in our earth's history?

First let's consider how one might change the radioactive decay rate, and then we will see whether such conditions existed in nature. Altering temperature from close to absolute zero to over 4,000 degrees Fahrenheit has failed to appreciably alter radioactive decay rates. High magnetic fields and high accelerations also have failed. However, high pressure has been found to change the decay rate for some isotopes.^{2,3} This can be explained by the fact that high pressures would squeeze electrons closer to the nucleus, increasing the likelihood of electron capture by the

²Brown W (2008) The origin of Earth's radioactivity. In *The Beginning: Compelling Evidence for Creation and the Flood, 8th Edition*. Center for Scientific Creation <<http://creationscience.com/onlinebook/Radioactivity2.html>> Accessed 2011 February 23

³ Liu L, Huh C (2000) Effect of pressure on the decay rate of ⁷Be, *Earth and Planetary Science Letters*, 180:163–167

nucleus. Electron capture results in the transformation of a proton into a neutron, which is the opposite of what happens in radioactive beta decay, when an electron is emitted by the atom.

If increasing the likelihood of electron capture can “reverse” beta decay, then what if we decreased the chance of electron capture? Could this increase the rate of decay? Indeed, the rate did increase under this condition in experiments performed by Dr. Fritz Bosch with the rhenium atom. He found the result to be a vast speedup of radioactive decay, by a factor of over a billion!⁴ What does this mean? If we used a radiometric dating method to analyze the sample that Dr. Bosch used and assumed the accepted “normal” decay rate, then we would think it took a billion times longer for the radioactive material in that sample to decay to the amount we measured instead of the short time it actually took.

Can you see the implication of this for the radiometric dating that has been used “across the board” in science? Dr. Bosch’s experiment implies the age of the material could be vastly younger than the measurement may seem to indicate. In fact, patents have actually been granted (e.g., US patent 5076971) for devices which utilize electrical devices to vastly speed up the decay rates of radioactive materials, for the purpose of decontaminating radioactive waste.⁵ The electrical influence on radioactive decay rates is consistent with the fact that free neutrons, that is, neutrons that are not confined to an atomic nucleus, have a half-life of only minutes, while the half-life of a neutron inside an atomic nucleus is millions of years. One obvious difference is that inside a nucleus, the neutron is closely surrounded by electrically charged particles.

Is there any time in earth’s history when there might have been electrical forces applied to the atoms of radioactive materials—electrical forces of sufficient energy that could have altered the decay rates of those radioactive materials? There is one such possibility; let us examine it now.

The Flood

According to the Hydroplate Theory², there was in earth’s history a time when there would have been sufficient electrical forces generated to alter the radioactive decay rates of earth’s radioactive materials. Here is the background needed for this explanation.

The piezoelectric effect

When quartz is placed under pressure, there is a voltage

⁴ Kerr RA (1999) Tweaking the clock of radioactive decay, *Science*, 286:882

⁵ Barker WA (1989) United States Patent 5076971, Method for Enhancing Alpha Decay in Radioactive Materials, Assignee: Altran Corporation, Sunnyvale, CA.

produced (Fig. 1, front page). Under tension, the opposite of compression, there is also a voltage produced with a reversed polarity. This inducement of voltage via pressure/tension is caused by the fact that, although the overall electrical charge of the quartz molecule may be neutral, there are regions of the quartz molecule that have a slightly more positive charge and other regions with a more negative charge. Forces that slightly deform the molecule move these regions, thereby moving the electrical charges present in those regions. Redistribution of electric charges results in changes of voltage across those regions.

This piezoelectric effect of quartz working through the pressure generated by earthquakes, has been known to produce electrical activity in granite, which is about 27% quartz. For example, during earthquakes, people have described the sky lighting up, ribbon-like flashes of lightning, globes of fire and sheets of flame. Also, iron has melted during an earthquake—an effect which could easily be due to the heating caused by electrical fields in the conductor, iron.⁶

The source of the (mechanical) force

So, what could possibly have happened in earth’s remote past to cause piezoelectric voltages, sufficient to alter radioactive decay rates? According to the Hydroplate theory, there were, before the Flood of Noah, large underground chambers of water. The crust above these water chambers was about 10 miles thick. Granite pillars supported the crust above the chambers. The quartz crystals in the granite of those pillars were aligned (due to the way the quartz crystals were formed, i.e., by extrusion and by the effect of tidal forces). Each crystal acted as a tiny battery, and since there was an alignment of the crystals, the result was the accumulation of electrical voltages (and thereby electrical currents and forces) resulting from each crystal’s piezoelectrically induced voltage changes.

As the crust broke open and the water surged up, leading to the great flood, the pressures involved in the breakup of the crust and the pressures in the pillars would be changing and be greatly increasing in cases of fracture, bending, and breaking of rock layers and granite pillars. This would have caused piezoelectrically generated electrical force inside the earth’s rock layers. If the electrical force was great enough, this could have altered the radioactive decay rate in those rock layers.

The undulating crust that is familiar to many in the case of earthquakes would be many times greater at this historic period when the earth’s crust broke open, and “the fountains of the great deep” were opened. This undulation of

⁶ The Sourcebook Project (1974) *Strange Phenomena A Sourcebook on Unusual Natural Phenomena*, Glen Arm, MD Vol. G1, and Vol. G2

the crust would result in tension and compression in the crustal rock. Evidence of this bending of the crust is found in lineaments, parallel linear structures of unknown origin, along which earthquakes tend to occur.⁷ The undulating crust would have caused the surface to fracture in just such parallel lines as these lineaments.

The electrons flowing through the rock would have encountered atoms of various sizes and types, resulting in acceleration/deceleration of the electrons. This acceleration produces bremsstrahlung radiation, which in turn produces free neutrons in surrounding material if the energy of the radiation is high enough.⁸ These neutrons could affect the rate of radioactive decay.

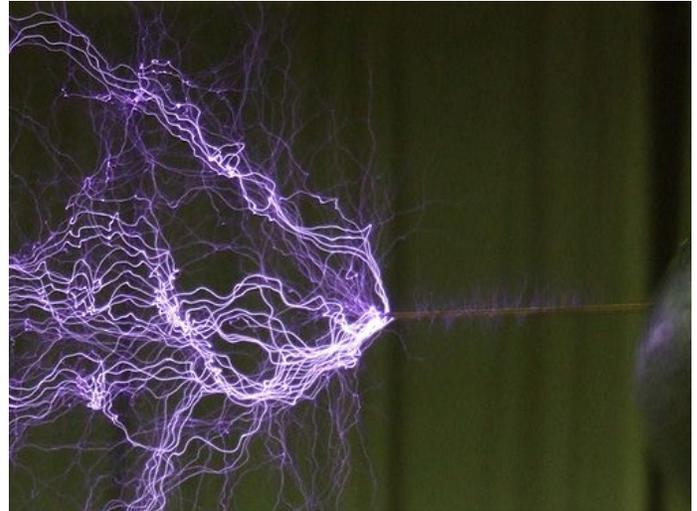
High energy electron flow inside solids can also produce heavy elements that will then decay into typical elements found in earth's crust. This has been observed in experiments performed in Kiev, Ukraine.^{9,10} The reported result of these experiments was the creation of isotopes of elements that are found in the earth's crust but which were not in the original material of the experiment. The energy output of some experiments was greater than the electrical energy input, and this, plus the appearance of newly formed elements, suggested that fission and/or fusion must have occurred to account for the energy. Also, the fission/fusion would have had to have occurred at an accelerated rate to account for the observed results.

At the Flood, electrical currents due to the mechanisms mentioned above could have vaporized the rock, and this would have led to vastly increased pressures in the rock. This pressure could then lead to more piezoelectric effects.

Another effect that could have been present during the Flood's crustal rupture is known as z-pinch (Fig. 2). The path taken by closely placed, nearly parallel, electrical currents are subject to a constricting effect, known as the z-pinch,¹¹ which would force nuclei closer together. Recall that pressure (i.e., forcing things closer together) has been

shown to affect decay rates, and the z-pinch would move particles into closer proximity to each other. Thus, it would seem that the z-pinch effect would contribute to the mechanisms involved in altering the radioactive decay rate of material.

Figure 2



Threads of light representing roughly parallel flows of electric current in a plasma. The electricity flowing through these parallel strands results in electric and magnetic fields which interact in a way that pulls, or squeezes, the strands of flowing electric current together. This same force, occurring in electricity flowing through rock, could have squeezed atoms together, resulting in fusion.

Summary

Today we have experimental evidence that radioactive decay rates can be altered. In fact, a patent has been granted for a method of decontaminating radioactive waste by accelerating the decay rate. Electrical forces are the key factor in these experiments. Today we also have experimental evidence that electrical forces can lead to the production of elements and isotopes, including radioactive ones.

The flood was a time of great physical stress in the earth's crust, in the opening of the "fountains of the great deep." The result of this stress was piezoelectrically generated electrical forces, which in turn could have altered decay rates through the mechanism elaborated above, leading to greatly accelerated decay. These events would cause today's methods of dating to give erroneously old dates. ❧

COMING EVENTS

Thursday, March 10, 7:00 P.M., Providence Baptist Church, 6339 Glenwood Ave., Raleigh, Room 631

Joe Spears will examine the tool known as radiometric dating. The ancient dates assigned to certain rocks and fossils have long been an issue for many creationists. We will look at recent research which sheds light on how this issue might be resolved.

⁷ Baeza AA, Zverev A, Malinnikov V (2004) Changes in geological faults associated with earthquakes detected by the lineament analysis of the aster (TERRA) satellite data. Pagina Web De Geofisica < http://www.geofisica.cl/papers/alonso2/Seismic_Risk.htm > Accessed 2011 Feb 28

⁸ Shkolnikov PL, Kaplan AE (1997) Laser-induced particle production and nuclear reactions, *J Nonlin Opt Phys Mater*, 6(2):161-167

⁹ Adamenko S, Selleri F, Van Der Merwe A (2007) *Controlled Nucleosynthesis: Breakthroughs in Experiment and Theory*, Springer Verlag, Dordrecht, The Netherlands

¹⁰ Proton-21 Electrodynamics Laboratory (2003) Results of experiments on collective nuclear reactions in superdense substance. <www.proton21.com.ua/articles/Booklet_en.pdf> Accessed 2011 February 23

¹¹ Bennett WH (1934) Magnetically self-focusing streams, *Physical Review*, 45: 890-897